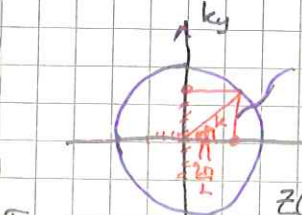
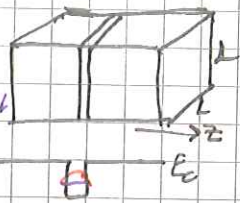


12. 2D-ellströngas

$\Psi_{nk} = e^{ik_x x} \cdot e^{ik_y y} \cdot \phi(z)$
 $E = \frac{\hbar^2}{2m} (k_x^2 + k_y^2) + E_z$

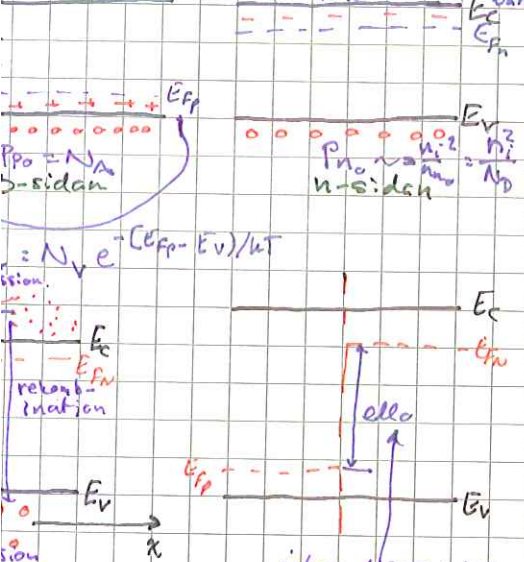


Fermienergin: E_F
 i) $N = n \cdot L^2 \int_0^{E_F} Z(E) \cdot dE$
 $n = \int_0^{E_F} Z(E) dE$
 $E_F = \frac{\hbar^2 k_F^2}{2m}$

$S(k) = \frac{\pi k^2}{(2\pi)^2} \cdot 2$
 $S(E) = \dots \frac{Z(E)}{Z(E)}$
 $Z(E)/L^2 = \frac{m}{\hbar^2 \pi}$
 $E_F = \frac{\hbar^2 k_F^2}{2m}$

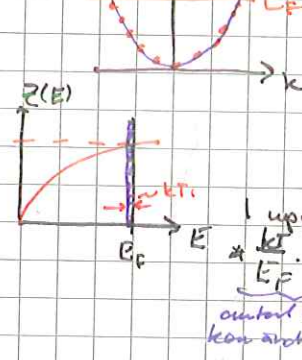
ii) $S(E) = \frac{2mE}{2\pi\hbar^2} L^2$. Om $E = E_F$, då är $S(E) = N$.

pn-övergång



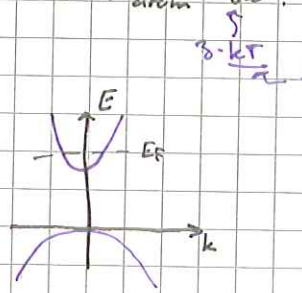
$\frac{2mE_F}{2\pi\hbar^2} L^2 = N = n L^2$

16.



$C_V = \frac{dE}{dT}$
 $E(T) = E(0) + \int_0^{E_F} Z(E) E dE$
 kompendiet: (appr. 1)
 $E(T) = E(0) + k_B T \cdot Z(E_F) \cdot k_B T = \frac{3}{2} k_B T \cdot \frac{N}{E_F} \cdot k_B T$
 antal e- som kan ändra energi.
 i uppgiften (appr. 2.)
 $E_F = k_B T \cdot \ln \left(\frac{n}{n_0} \right)$
 antal som kan ändra energi.

17) atomära vibrationer



$E(T) = N_{atom} \cdot 3k_B T$
 $E = \frac{3}{2} N k_B T$
 $C_V = \frac{dE}{dT} = \frac{3}{2} N k_B$
 $E = \frac{3}{2} N k_B T$
 $C_V = 3N k_B$

